Precursor stoichiometry in CH₃NH₃PbBr₃. Structure-property relationship and LED implication

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Organic-inorganic cubic perovskites





L- Gil-Escrig, G. Longo, A. Pertegas, C. Roldan-Carmona, A. Soriano, M. Sessolo and H. J. Bolink, *Chem. Commun*, 2015, **51** K. Tvingstedt, O. Malinkiewicz, A. Baumann, C. Deibel, H. J. Snaith, V. Dyakonov and H. J. Bolink, *Sci. Rep.*, 2014, **4**.

Analyzed samples

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Stoichiometric

Non-stoichiometric

 $CH_3NH_3:PbBr_2 = 1$ $CH_3NH_3:PbBr_2 = 3$

Single step deposition technique



Powder diffraction analysis

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Polycristalline powders precipitated from concentrated solutions

Precipitating agent: chlorobenzene

Dryed at 75° for 48 hours, and then annealed at 90°C for 3h. Sample 3:1 was further annealed at 115°C for 24 hours

Solid state C¹³ NMR analysis





freeMA/Pb-MA= 1.38 Even after thermal treatment (115°C 24h), the same ratio between bounded and unbounded methilammonium is kept treatment (90°C 3 hours)

Device structure

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3-1 OLED device



1-1 OLED device



Morphological study

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MA/Pb =3



Relationship of performances and structure



Optical behavior





Electroluminescence and photoluminescence

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Excitation wavelength: 365nm

Trap states filling



No photoluminescence

Photoluminescence at 527 nm

Trap assisted recombination



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Conclusions

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- Precursor stoichiometry of CH₃NH₃PbBr₃ have a fundamental role in the optoelectronic properties and in film formation
- An excess of methylammonium affects the perovskite bandgap
- An excess of methylammonium bromide favors the surface coverage and the creation of a compact and uniform perovskite layer, suitable for optoelectronic application
- The excess of MABr passivates the trapping states present on the crystals allowing photoluminescence even at low excitation intensities
- The electroluminescence in the device likely derives from trap assisted recombination processes.

Acknowledgments

Dr. Henk J. Bolink Dr. Michele Sessolo Dr. Maria Monrabal-Capilla Lidon Gil-Escrig Laura Martinez-Sarti Mª Cristina Momblona Rincor David Forgacs Enrico Bandiello Antonio Pertegas-Ojeda Jorge Pablo Avila-Gomez Jorge Ferrando-Garcia Araceli Miquel-Sempere







Thank you for your attention!

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